organic papers

Received 26 September 2006 Accepted 1 November 2006

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 292 K Mean σ (C–C) = 0.007 Å R factor = 0.053 wR factor = 0.118 Data-to-parameter ratio = 7.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Two tautomers in one crystal structure: ethyl 6(4)-oxo-1,6(1,4)-dihydropyrimidine-5-carboxylate

The crystal structure of the title compound, $C_7H_8N_2O_3$, contains a 1:1 mixture of two tautomers, *viz*. ethyl 6-oxo-1,6-dihydropyrimidine-5-carboxylate and ethyl 4-oxo-1,4-dihydropyrimidine-5-carboxylate. The molecules are linked into a three-dimensional network by N-H···O, C-H···O and C-H···N hydrogen bonds.

Comment

Pyrimidine-containing compounds have been intensively studied because of the biological activity which they usually exhibit (Rosowsky *et al.*, 2003). The title compound, (I), which may be a novel precursor of bioactive heterocyclic compounds, was designed and synthesized in our laboratory.



Interestingly, the crystal structure (Fig. 1) consists of a 1:1 mixture of two tautomers, involving the exchange of a H atom between the two ring N atoms. The ¹H and ¹³C NMR data for (I) suggested that only one tautomer is present in solution. A similar case of prototropic annular tautomerism has been studied in an imidazole derivative (Kubicki, 2004).

All bond lengths and angles are within normal ranges. The crystal packing (Fig. 2) of (I) is stabilized by strong $N-H\cdots O$ hydrogen bonds and weak $C-H\cdots O$ and $C-H\cdots N$ hydrogen bonds between different tautomers.



Figure 1

The asymmetric unit of the title compound. Displacement ellipsoids are drawn at the 50% probability level. H atoms are depicted as spheres with arbitrary radii.

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Experimental

A solution containing diethyl 2-(ethoxymethylene)malonate (1.08 g, 1 mmol) and formamidine (0.264 g, 6 mmol) in anhydrous ethanol (30 ml) was refluxed for 24 h; the solvent was then removed under reduced pressure. Crystals suitable for an X-ray investigation were obtained by recrystallization from methanol and chloroform (1:2 ν/ν).

Z = 8

 $D_x = 1.441 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

 $\mu = 0.12 \text{ mm}^{-1}$

T = 292 (2) K

Plate, colorless

 $R_{\rm int} = 0.053$

 $\theta_{\rm max} = 26.0^{\circ}$

 $0.40 \times 0.10 \times 0.03~\text{mm}$

1526 independent reflections

1121 reflections with $I > 2\sigma(I)$

Crystal data

 $\begin{array}{l} C_{7}H_{8}N_{2}O_{3}\\ M_{r}=168.15\\ \text{Monoclinic, }Cc\\ a=4.768 \ (3) \text{ Å}\\ b=25.590 \ (16) \text{ Å}\\ c=12.821 \ (8) \text{ Å}\\ \beta=97.687 \ (10)^{\circ}\\ V=1550.4 \ (17) \text{ Å}^{3} \end{array}$

Data collection

Bruker SMART APEX CCD diffractometer φ and ω scans Absorption correction: none 4986 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.053$ $wR(F^2) = 0.118$ S = 1.031526 reflections 219 parameters H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0587P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.25 \text{ e } \text{ Å}_{-3}^{-3}$

 $\Delta \rho_{\rm min} = -0.17 \text{ e} \text{ Å}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
N1-H1···O4 ⁱ	0.86	2.00	2.847 (5)	170
$C3-H3 \cdot \cdot \cdot N3^i$	0.93	2.54	3.188 (7)	127
$N4-H4A\cdotsO1^{ii}$	0.86	2.01	2.789 (5)	151
$C3-H3\cdots O5^{ii}$	0.93	2.55	3.307 (6)	139
C10−H10···O4 ⁱⁱ	0.93	2.50	3.135 (6)	125
$C10-H10\cdots O5^{iii}$	0.93	2.55	3.281 (7)	136

Symmetry codes: (i) x + 1, y, z; (ii) $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$; (iii) $x - \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$.

All carbon-bound H atoms were positioned geometrically and then treated as riding, with $Csp^2-H = 0.93$ Å, methyl C-H = 0.96 Å and methylene C-H = 0.97 Å, and with $U_{iso}(H) = xU_{eq}(C)$, where x =1.5 for methyl and 1.2 for other H atoms. Atoms H1 and H4A, attached to N1 and N4, respectively, were located in a difference map and refined as riding in their as-found positions, with N-H = 0.86 Å and $U_{iso}(H) = 1.2U_{eq}(N)$. In the absence of significant anomalous scattering effects, 1391 Friedel pairs were merged.



Figure 2

A view of the packing in the crystal structure. Dashed lines indicate hydrogen bonds. H atoms not involved in hydrogen bonding have been omitted.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

The authors acknowledge Dr Guodong Yin for fruitful discussions.

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